

(04 Marks)

USN

First/Second Semester B.E. Degree Examination, June/July 2018 Engineering Physics

	-	Engineering Physics	
Tin	ne: 3	hrs. Max. Marks:10	0
Not		. Answer any FIVE full questions, choosing at least two from each part. Physical constants: Planck's constant $h = 6.63 \times 10^{-34} \text{ J-S}$, Electron charge $e = 1.6 \times 10^{-19} \text{ C}$, Electron mass $m = 9.11 \times 10^{-31} \text{kg}$, Velocity of light $C = 3 \times 10^8$ m/s, Neutron mass $= 1.675 \times 10^{-27}$ kg, $\epsilon_0 = 8.85 \times 10^{-12}$ F m $K = 1.38 \times 10^{-23}$ J/K	
		PART - A	
1	a.	Choose the correct answers for the following: i) According to Wein's law, the wavelength of maximum intensity λ_m is (04 Mar	ks)
		A) $\lambda_m \alpha T$ B) $\lambda_m \alpha \frac{1}{T}$ C) $\lambda_m \alpha T^4$ D) $\lambda_m \alpha \frac{1}{T^4}$	
	c.	ii) de Broglie wave length of an electron accelerated by a potential of 100V is A) 0.0012nm B) 0.1226 nm C) 1.226 nm D) 12.26 nm iii) The wave nature associated with electrons in motion was verified by A) Photoelectric effect B) Compton effect C) Raman effect D) Diffraction by Crystals iv) The momentum of a free particle carrying energy E and mass m is A) 2mE B) √2mE C) 2√mE D) m² E². What is Planck's radiation law? Show how Wein's law and Rayleigh − Jean's law can derived from it. (06 Mar What is Matter wave? Derive an expression for de − Broglie wavelength using grovelocity concept. (06 Mar Find the energy of the neutron i.e eV whose de − Broglie wavelength is 1A³. (04 Mar	ks) oup ks)
2	a.	Choose the correct answers for the following: i) The Product of Uncertainties between position and momentum is given by	ks)
		A) $\Delta x \Delta p \ge \lambda$ B) $\Delta x \Delta p \ge \frac{h}{2}$ C) $\Delta x \Delta p \ge mv$ D) $\Delta x \Delta p \ge \frac{h}{4\pi}$	
		The probability of locating a particle is maximum A) within the wave packet B) at the nodes of the wave packet C) cannot be determined D) none of these iii) The energy corresponding to the first permitted energy level for a particle in an infinite potential well is called A) excited state B) zero point energy C) meta stable state energy D) none of these iv) If an electron moves in one dimensional box of length 2nm, the normalization constatis	
		A) $1(nm)^{-1}$ B) $2(nm)^{-1}$ C) $\sqrt{2}(nm)^{-1}$ D) zero	
	D.	Set up time independent Schrodinger wave equation. (06 Mar. Show that electrons cannot exist in the nucleus of an atom. (06 Mar.	

ground state and first excited state.

d. An electron is bound in one dimensional box of width 0.16nm. Find the energy values in the



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3	a. Choose the correct an	swers for the follow	ing:	(04	Marks)			
	i) Specific heat of 1	mole of electron gas	under constant volum	e is				
	A) $\frac{3}{5}$ R ii) The expression of	B) $\frac{R}{2}$	C) 3R	D) $\frac{3}{2}$ R				
	ii) The expression of	electrical resistivity		As				
	A) $\frac{ne^2\tau}{m}$	B) $\frac{m}{ne^2\tau}$	C) $\sqrt{\frac{ne^2\tau}{m}}$	D) $\sqrt{\frac{m}{ne^2\tau}}$				
	iii) The fermi factor		is					
	A) 1	B) $\frac{4}{2}$	C) 0	D) 2				
	iv) According to qua							
			C) overlapping		.1.			
	b. Using the free electron	n theory, derive an e	expression for electrica		Marks)			
	c. Discuss the probabili	ty of occupation of	various energy states					
	$T > 0^{\circ}$ K on the basis of	of fermi factor	various energy states		Marks)			
	d. Find the temperature		0% probability that a					
	above fermi energy w		.oro producting that a		Marks)			
		and the state of the state of						
4	a. Choose the correct an	swers for the follow	ing:	(04	Marks)			
	i) Electronic Polariz							
	A) Independen	t of temperature	B) Increases w	rith temperature				
	C) Decreases v	vith temperature	D) None of the	se				
	ii) For Ferromagnet	ic substance, the Cui	rie – Weiss law is give	n by				
	ii) For Ferromagnetic A) $X = \frac{C}{T}$	B) $X = C$	C) $X = (T - \theta)$	D) $X = C$				
)			
	iii) The Polarisation							
			C) orientation					
	iv) Sulphur is an elemental solid dielectric of atomic weight 32.07 and density 2.07 ×10 ³							
	kg/m ³ . The number of atoms per unit volume for sulphur is							
	A) 3.89×10^{28} /m ³ B) 3.89×10^{25} /m ³ C) 9.3×10^{24} /m ³ D) None of these b. What is Internal field? Derive an expression for internal field in case of one dimensional							
			ssion for internal field		Marks)			
	array of atoms in diele				(Marks)			
			os Polorizability 7 ×					
	d. An elemented solid dielectric material has Polarizability 7×10^{-40} F $-$ m ² . Assuming the internal field to be Lorentz field. Calculate the dielectric constant for the material, if it has							
	3×10^{28} atoms/m ³ .	ficine neig. Culcul	are the dictional comm		(Marks)			
	Jan Monto, III.							
			PART - B					
5	a. Choose the correct an			(04	Marks)			
		on in diode laser is						
	A) Optical pur		B) Electric dis-	charge				
	C) Reverse bias		D) Forward bia	ıs				
	ii) The distribution o	f number of atoms i	s different discrete ene					
	A) Fermi – Dira	e distribution		Boltzmann distributi	on			
	C) Bose - Einst	ein distribution	D) None of th	ese				
		he meta stable is abo	$\frac{\text{out}}{\text{C) }10^2} \sec$	75				
	A) 10^{-3}	B) 10 ⁻¹³	C) 10 ²	D) 10 ⁻⁹				



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	iv) Image is stored on a hologram in the form						
		Diffraction pa	ttern				
		None of these					
	Explain the terms spontaneous emission and stin		(04 Marks)				
C.	Describe the construction of He – Ne laser and	explain its workin					
	diagrams.		(08 Marks)				
d.		iw per pulse and p	oulse duration is 20ns. The				
	number of photons emitted per pulse is estimate	ed to be 1.04/ × 1					
	the emitted laser.		(04 Marks)				
	Classification of the following in		(04 Marks)				
a.	그는 사람들에게 어디지면 아이들이 되는 것도 할 것이 될 것이 없는데 없는데 얼마나 되었다. 그는데 나는 아니는 이 모든데 모든데 나를 했다.		(04 (viaiks)				
	i) The superconductor behaves like a Perfect	C) Diamagnet	D) None of these				
	A) Paramagnet B) Ferro magnet	C) Diamagnet	D) None of these				
	ii) The critical temperature of mercury is A) 4.2 K B) 2.4 K	C) 6.2 K	D) 7.8 K				
	iii) The quantum of magnetic flux is given by		2) 7.01				
		co h	D) 2πh				
	A) $\frac{2h}{e}$ B) $\frac{h}{2e}$	C) $\frac{h}{2\pi e}$	D) $\frac{2\pi h}{e}$				
	iv) The attenuation of a fiber – optic cable is e						
		C) decibel/km	D) joule/km				
Ь.	Define Super conductivity and explain Type - I	and Type - II sup	er conductors. (06 Marks)				
C.	. Describe different types of optical fibres, with n	eat diagrams.	(86 Marks)				
d.	. The attenuation of light in an optical fiber is	3.6 dB/km. What	t fractional initial intensity				
	remains after 1km?		(04 Marks)				
a.	. Choose the correct answers for the following:	. Win 200 200	(04 Marks)				
	i) The relation between atomic radius and lattic	e constant in FCC	structure is				
	A) $a = 2r$ B) $a = 2\sqrt{2} r$	C) $a = \sqrt{3}r$	D) $a = \frac{4r}{r}$				
	ii) Which of the following crystal structure is h	aving the least coo	ordination number?				
		B) Body centere					
	C) Face centered cubic	D) None of these					
	iii) The inter planar spacing in a crystal is 1A° and the glancing angle is 35°. For the first						
	order Bragg reflection to take place, the wa	velength of X – ra	ys is D) 0.819A ⁰				
	A) 1.147A ⁰ B) 0.573A ⁰	C) 1.638A					
	iv) The inter atomic distance between the sodii A) 5.51A ⁰ B) 5.62A ⁰	im and chlorine at $6.62A^0$	D) 2.81 A ⁰				
h	Derive Bragg's law for X - ray diffraction in cr		(04 Marks)				
c.	Define Coordination number and atomic packir	Define Coordination number and atomic packing factor. Calculate the packing factor for SC,					
	BCC and FCC structures.		(08 Marks)				
d.	. Copper has FCC structure and the atomic rac	lius is 0.1278nm.	Calculate the inter planor				
	spacing for (111) and (321) planes.		(04 Marks)				



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(08 Marks)

(04 Marks) a. Choose the correct answers for the following: i) In a carbon nanotube, the bond between the carbon atom is A) metalic C) hydrogen D) covalent B) ionic ii) The Ultrasonic waves are sound waves having B) Velocity less than 330 ms A) Velocity greater than 330 ms D) Frequency less than 20 KHz C) Frequency greater than 20 KHz iii) The ultrasonic waves are produced by B) Electric tuning fork A) Electromagnetic induction C) piezo electric effect D) Inverse piezo electric effect iv) A constant testing of product without causing any damage is called A) minute testing B) destructive testing C) non - destructive testing D) random testing b. Explain Carbon nanotubes and its application by giving their physical properties. (08 Marks)

c. Describe a method of measuring velocity of ultrasonic waves in liquids.
